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1. aromatic ring accepts a pair of electrons from a nucleophile to make an anionic intermediate. Must have a strong EWG, like a nitro (NO₂) group to stabilize the anion. 2. Elimination of a leaving group. H⁻ is a very poor leaving group, so the nucleophile must add at the site of a good leaving group. Often it is a halogen.

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And you have a methyl group: one carbon attached to the fourth carbon of our main backbone. So this is going to be four methyl, this is our methyl group right here, 4-Methylnonane. So it was brought up, I think the user name is Minoctu, and they correctly corrected me, that there would never be such a thing as 2-Propylheptane.

Correction - 2-propylheptane should never be the name ...

2.2 Acyl Substitution: Nucleophilic Addition-Elimination at the Acyl Carbon An acyl substitution can occur as follows but always requires a leaving group at the carbonyl carbon: The order of relative reactivity of acyl compounds goes as follows: acyl chloride > acid anhydride > ester > amide

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Alcohol Nomenclature

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- Step 2: Disconnect one of the groups attached to this carbon. - What remains are on the carbonyl compound that can be used to prepare the alcohol. The detached part will be on the Grignard reagent.

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